

## SBI4U Review

### Exam Format

	# Questions	Marks
Multiple Choice	55	55
Matching	15	15
Short Response	9	40
		<b>110</b>

Duration: 2.5 hours

Weight: 30%

- Starts at 9 AM
- Arrive 8:45
- Full uniform
- Schoolbags at side of room
- Bring textbook
- At desk: textbook, pencil/eraser/pen
- No phones
- No calculators

Appreciating biology is about identifying the **enduring understandings**.

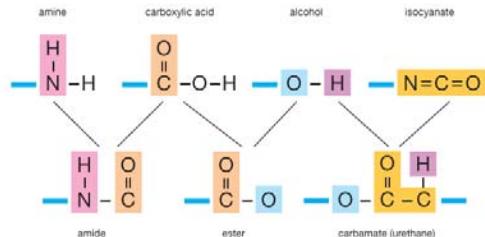
- Enduring Understandings = Concepts and Big Ideas
  - Examples vs. “Facts”
  - Fundamental ideas that can enhance content understanding from all units (even from last year!)
- Cross-curricular connections and themes
  - Be able to identify examples from as many units as possible

Enduring Understanding	Specific Details	Examples
A molecule's polarity influences the way it interacts with other molecules.	<ul style="list-style-type: none"> <li>• Ionic charges, or polar functional groups make a molecule hydrophilic.</li> <li>• Hydrocarbons are hydrophobic.</li> <li>• “Like dissolves like”.</li> </ul>	<ul style="list-style-type: none"> <li>• Lipids are insoluble in water.</li> <li>• The cell membrane is mostly hydrophobic.</li> <li>• Side chains of amino acids determine AA properties.</li> <li>• Ions and polar molecules cross channel proteins</li> <li>• Steroid vs. protein hormones</li> </ul>

### Unit 1: Biochemistry

- Basic molecular properties
- Biochemical reactions
- Macromolecules
- Membranes & Transport

### Functional groups in monomers and polymers



### Functional groups

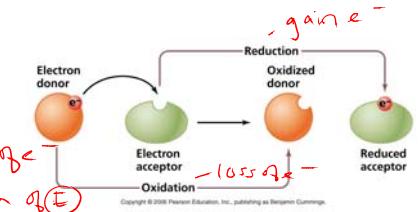
- Physical properties
- Chemical properties

### Redox

- Specific examples?

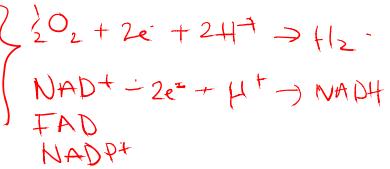
- Energy changes

$$\text{Reduction} = \text{gain } e^- + \text{gain } q(E)$$



cell resp

photosynthesis

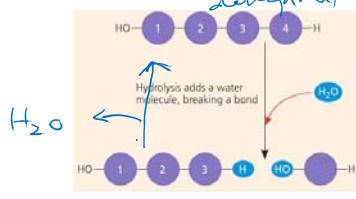


### Hydrolysis

- Specific examples?
- Energy changes

- polymerization
- anabolism
- $(\text{glucose})_n \rightarrow \text{glycogen} \Rightarrow \text{energy}$

### / condensation / dehydrogenation

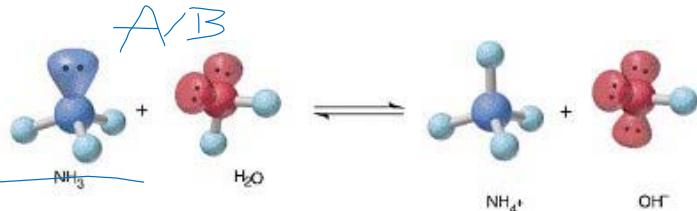
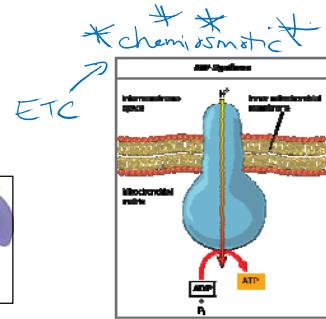
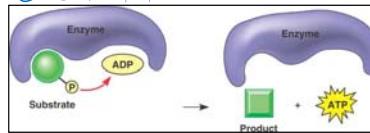


### Phosphorylation

- Specific examples?

- Energy changes -  $\textcircled{E}$  storage

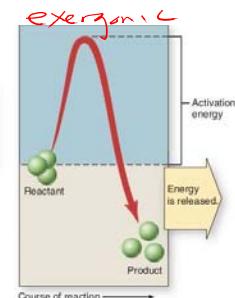
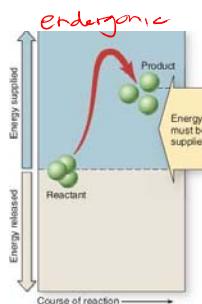
### Substrate level



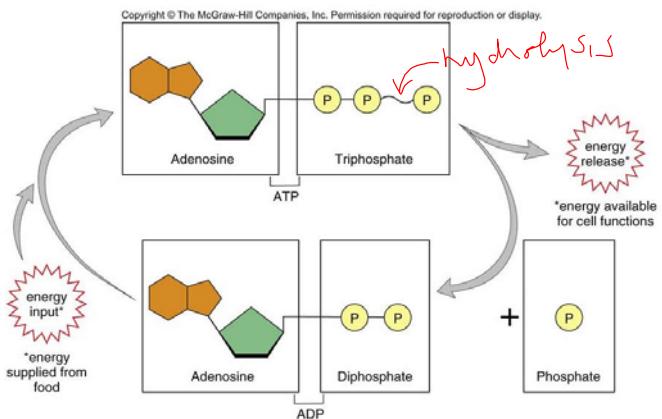
- Significance in the body?

Optimal pH  
↑ proteins

### Energy changes in reactions

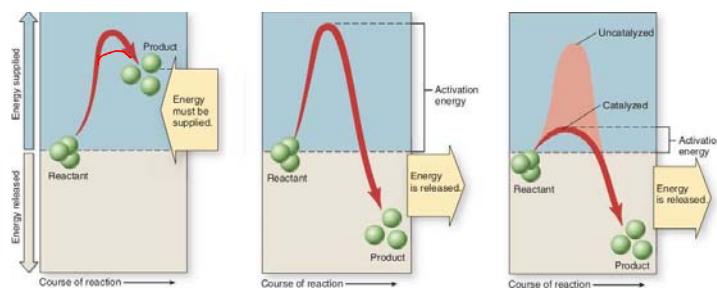
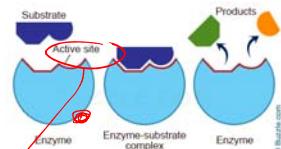


- Activation energy?

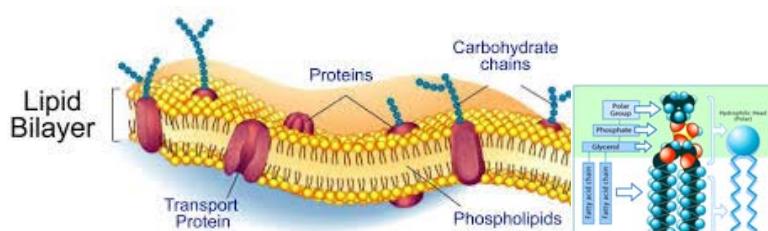


## Enzymes

- function - speeds it up
- significance - allows rxns at body temp
- Importance of shape
  - allosteric inhibition (operons)
  - competitive inhibition (eg/mutisc + O<sub>2</sub>)



Monomer	Polymer	Cellular structure
Amino Acid	Protein	Intermediate filament
	DNA strand	Chromosome
Nucleotide		
Monosaccharide	Starch	Starch grains in a chloroplast
Fatty acid	Fat molecule	Adipose cells with fat droplets



- “Fluid Mosaic”?
- important properties

## Examples?

- metabolism
- nervous system
- excretory system

